

THE SIMULATION OF THE TIME CHARACTERISTICS OF THE PHOTORESISTIVE STRUCTURES BASED ON THE CADMIUM LEAD SULFIDES

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ABSTRACT

The structures of the lead sulphide and lead selenide based photoresistors are examined. The simulation allowed us to explain the experimentally observed differences in the photoconductivity relaxation time between two different types of photoresistors.

Keywords: photoresistor, lead chalcogenides, $\text{Pb}_x\text{Cd}_y\text{S}$ -based structures, carrier lifetime, relaxation time

1. INTRODUCTION

Photoresistors (PRs) on the base of lead chalcogenides (PbS, PbSe, PbTe) have found widespread applications in infrared (1-5 microns) impulse optical-electronic systems (OES) since the 1930s. In order to produce the first PRs, monocrystalline or epitaxial layers were used (Ravich, Efimova, and Smirnov 1968). However, the OES developers encountered two problems then. The first problem was that the layers with a low dark carrier concentration (preferably less than 10^{16} cm^{-3}) were required to increase the signal, but the technology of the pure layers was too expensive. The second problem was that in order to increase the sensitivity of PRs, the lifetime must be increased, but in extremely pure PbS crystals the value of τ did not exceed 10 μs .

A special feature of the impulse OES is the fact that the main criterion for the PR quality is not the increment of conductivity $\Delta\sigma$ under the influence of radiation, but the ratio $\Delta\sigma/\sigma_0$ divided by the effective radiation power at the fixed modulation frequency F_e . Here σ_0 is the dark conductivity of PR. Therefore, while the values τ and σ_0 are being simultaneously changed, their ratio remains constant.

It was believed that the recombination processes of the nonequilibrium carriers produced by IR radiation occur mainly according to the Auger process, namely:

$$\tau p^2 = (2 \dots 3) \cdot 10^{27} \text{ cm}^{-6} \cdot \text{s} \quad \text{for PbS} \quad \text{and} \\ \tau n^2 = 4 \cdot 10^{27} \text{ cm}^{-6} \cdot \text{s} \quad \text{for PbSe (Baryshev 2000).}$$

The carrier lifetime in a number of the bulk PbS crystals was measured by T.S. Moss (1953) who managed to obtain the values for the lifetime in the range from $6 \cdot 10^{-10}$ to $6 \cdot 10^{-6}$ s. In the early 1950s, R.A. Smith (1954) showed that the lifetime of one type of the carrier (electrons or holes) can be extended, if the carriers of another type are localized inside a solid or on its surface. Later, A. Rose (1963) introduced the concept of “attachment level”. Creating the system consisting of small crystallites, for which the surface-to-volume ratio is much higher than the one for a single crystal, was a good idea. Therefore, presently there are a large number of PbS- and PbSe-based PR-making technologies applying the vacuum deposition and the chemical deposition of layers (Johnson 1984, Saloniemi 2000).

The purpose of this paper is to show the possibility of the simulation of the carrier lifetime in polycrystalline photosensitive PbSe- and Pb-Cd-S-based layers.

It should be noted that the analysis of the PR parameters based on the lead chalcogenides demonstrates the following regularities: despite of the differences in the energy-gap widths ($E_{g \text{ PbS}} = 0.4 \text{ eV}$, $E_{g \text{ PbSe}} = 0.25 \text{ eV}$

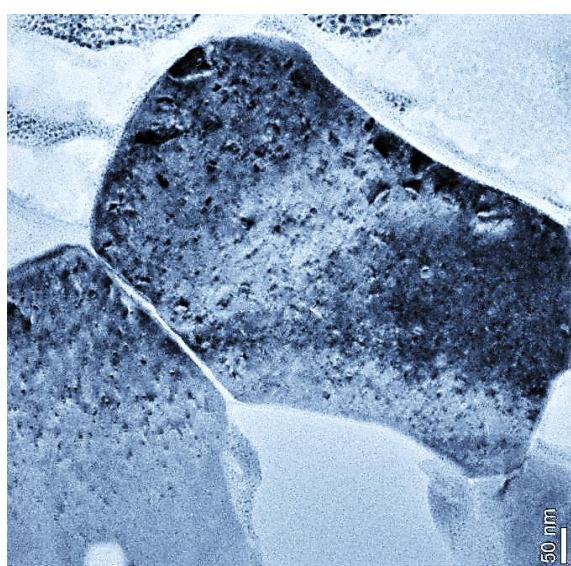
at room temperature), the values of the voltage sensitivity are quite close, while the effective values of the photoconduction relaxation constant differ by 20 times. Namely, there are 15 μs for PbSe and 200...450 μs for PbS. The doping of the PbS films by halogens (I, Br) leads to decreasing the time constant of the PbS-based structures up to 15 μs or less (Maskaeva 2004).

The integration of the lead and cadmium sulfides, which are the substitutional solid solutions ($E_{g \text{ CdS}} = 2.42 \text{ eV}$ at room temperature), in various proportions allows to conduct broad changes in the energy-gap width of the $\text{Pb}_x\text{Cd}_y\text{S}$ material. These polycrystalline films are obtained by the hydrochemical deposition from the aqueous solutions (Markov, Maskaeva, and Ivanov 2006) as well as by the thermal evaporation

(Aleksandrova, Maksimov, Moshnikov and Chesnakova 2008).

2. THE PHOTOSENSITIVE FILM STRUCTURE

In the case of substitutional solid solutions, the prepared films have varying composition in thickness. The solid solution based on the lead sulfide is formed at the depth of up to 40 nm approximately, while the solid solution based on the cadmium sulfide is formed at the depth of from 70 to 200 nm. In addition, there is a transition region with a thickness of 40-70 nm (Forostyanaya, 2015). During the transition from the PbS to the $Cd_x Pb_{1-x} S$ the conductivity type changes from “*n*” to “*p*”, while the carrier concentration decreases by 3-5 orders of magnitude with the time constant τ decreasing up to 15-60 μs .



a)



b)

Figure 1: The microphotographs of the cross-section of the physically (a) and chemically (b) deposited PbS-based photoresistors obtained with the help of the high-resolution transmission electron microscopy

Due to the low carrier mobility in polycrystalline films ($1 \text{ cm}^2/(\text{V}\cdot\text{s})$ as it is stated by Petritz (1958)), the sensitivity of PR can be characterized by the ratio τ/p_0 where p_0 is the dark carrier concentration.

The desire to increase the sensitivity of devices by increasing the carrier lifetime and to decrease the dark conductivity σ_0 , led to the development process of PR “sensitizing”. For this purpose, in the case of the PbS-based structures, oxygen was introduced into the polycrystalline film (Butkevich, Globus and Zalevskaya 1999), that is, either the high-temperature annealing in case of the physical deposition or the addition of the additives in the process of the chemical deposition, while for the PbSe-based structures, oxygen, sulfur, selenium and halogens were used (Humphrey and Scanlon 1957).

The process of conduction in the polycrystalline layers of the lead chalcogenides is greatly influenced by the intergranular barriers as well as by grain structure and size. Therefore, the study of the film morphology is of a particular importance. In Figs. 1 (Mohamed, Abdel-Hafiez, Miroshnikov, Barinov and Miroshnikova 2014) and 2 (Miroshnikova, Miroshnikov, Presnyakov and Mohamed 2018), the microphotographs are presented of the cross-section of the PbS- and PbSe-based structures, respectively, obtained with the help of the high-resolution transmission electron microscopy (HRTEM – STEM/TEM Titan 80-300). In these photographs, the light keys belong to the light elements (primarily, oxygen), the dark color shows the atoms with the big numbers (sulfur, selenium, lead and iodine). It is known (Maraeva, Moshnikov and Tairov 2013) that iodine being a catalyst provides a high sensitivity for the PbSe-based PRs due to the penetration of oxygen into the grain, which originally had *n*-type conductivity.



Figure 2: The microphotograph of the cross-section of the chemically deposited PbSe-based photoresistor obtained with the help of the high-resolution transmission electron microscopy

In contrast with the PbSe-based film, in the PbS-based film one can clearly see the oxygen-containing impurities (the term introduced by Butkevich, Globus and Zalevskaya (1999)) at the crystallite boundaries. The introduction of oxygen into the PbS structure leads to the formation of acceptor states in the forbidden band. This, in turn, increases the hole lifetime up to 600 μs , as there takes place the change of the conductivity from n - to p -type. The dark conductivity σ_0 will also increase in this case.

Based on the analysis of the PR structure, one can presuppose that there are two mechanisms for the overcompensation of the n -type conductivity in the lead chalcogenides. Namely, in PbS, the acceptor levels are formed both inside the grain (due to the oxygen diffusion) and on their surface (due to the oxygen-containing impurities), while for PbSe, PbCdS the enhanced oxygen diffusion mainly occurs with iodine and bromine acting as the catalysts for this process. Oxygen as an isoelectron impurity replaces sulfur, while in the PbCdS substitutional solid solution it can have a low concentration of 10^{18} cm^{-3} , approximately.

3. SIMULATION OF THE PHOTORESISTIVE STRUCTURES RELAXATION TIME

The photoresistive structures relaxation time is determined by the majority-carrier lifetime. As it was shown by Miroshnikova, Miroshnikov, Presnyakov, Mohamed (2018), in PbS, the concentration of holes released due to the electron trapping by the oxygen-containing impurities increases linearly with the number of the oxygen-containing impurities (the dependence $p_2(\tau)$ in Fig. 3a) by the law $p_2 = 1.4 \cdot 10^{20} \tau$, while the recombinations by the Auger process leads to the carrier concentration decreasing.

The influence of iodine in PbSe manifests itself in the stronger dependence $p_2 = 2 \cdot 10^{21} \tau$ (Fig. 3b), but in this case there also occurs the overcompensation of the conductivity type ($n - p_2$ curve in Fig. 3b). As a result, one has a very bounded area of the acceptable values of the carrier lifetime in PbSe that provide the low dark carrier concentration.

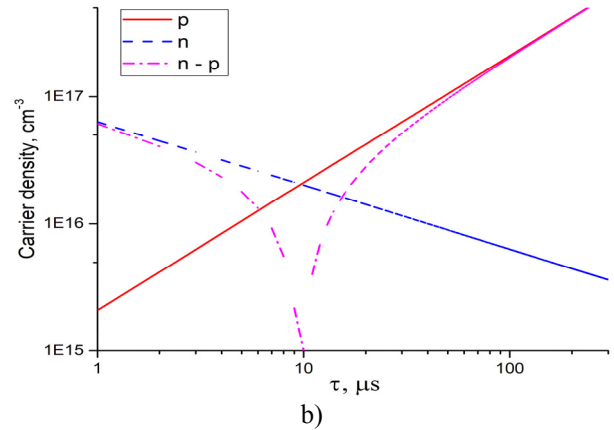
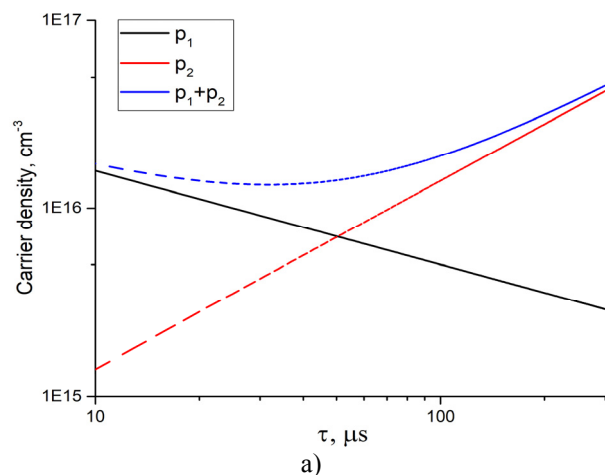


Figure 3: The time dependences of the carrier concentration in the structures based on PbS (a) and PbSe (b)

In $\text{Cd}_x\text{Pb}_{1-x}\text{S}$ substitutional solid solution, the oxygen adsorbed by the surface is of a great significance. Our studies using the EDX and AES methods have shown that oxygen is fixed on the surface in the area of lead sulfide. Adsorbed oxygen also creates acceptor traps decreasing the dark resistance.

4. CONCLUSION

Thus, it has been shown that in the case of PbS, there occurs a large variation of τ values acceptable for making the high sensitivity PRs. At the same time, in the case of PbSe and $\text{Cd}_x\text{Pb}_{1-x}\text{S}$ substitutional solid solution the area of the acceptable values of the carrier lifetime becomes very bounded due to the superposition of the two processes: the overcompensation of the conductivity polarity and the high rate of the oxygen diffusion deep into the microcrystallites.

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